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Chapter 16 Thermal Energy And

Terms in this set (...) Heat. the transfer of thermal energy from one object to another because of a difference in temperature. Temperature. a measure of how hot or cold an object is compared to a reference point. Absolute Zero. a temperate of 0 kelvins. Thermal

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2. water cools as it transfers thermal energy to the room, and the cool water returns to the boiler to be reheated 3. a thermostat controls the temperature, turning on the hot water flow if the room is too cold and off the hot water flow if

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the room is the desired temperature

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The energy needed to raise the temperature of 1 gram of water by 1 degree Celsius Chapter 16 Thermal Energy and Heat Vocabulary - Russo, Christopher 20 terms Christopher_Russo

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Chapter 16 Thermal Energy and Heat
Section 16.1 Thermal Energy and Matter (pages 474-478) This section defines heat and describes how work, temperature, and thermal energy are related to heat. Thermal expansion and

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contraction of materials is discussed, and uses of a calorimeter are explained. Reading Strategy (page 474)

Chapter 16 Thermal Energy and Heat Section 16.1 Thermal ...

the amount of heat or energy required to raise the temperature of 1 gram of a substance by 1 degree celsius.

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□ Thermal energy depends on mass, temperature, and phase (solid, liquid or gas) of an object. Thermal Contraction and Expansion. □ Thermal expansion is an increase in the volume of a material due to a temperature increase. □ Thermal

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expansion occurs when particles of matter move farther apart as temperature increases.

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The transfer of thermal energy from one object to another because of a temperature difference. The measure of

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how hot or cold something is compared to a reference point, related to average kinetic energy. Defined as having a temperature of 0 Kelvin. Occurs when particles of matter move farther apart as temperature increases.

Physical Science-Chapter 16 Thermal Energy and Matter ...

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concluded that heat is not a form of matter. 474 Chapter 16 FOCUS Objectives 16.1.1 Explain how heat and work transfer energy. 16.1.2 Relate thermal energy to the motion of particles that make up a material. 16.1.3 Relate temperature to thermal energy and to thermal expansion. 16.1.4 Calculate thermal energy, temperature

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change, or mass

Section 16.1 16.1 Thermal Energy and Matter

The Thermal Energy and Heat chapter of this Prentice Hall Physical Science Companion Course helps students learn the essential physical science lessons of thermal energy and heat. Each of these

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Chapter 16 - Thermal Energy

Chapter 16 Thermal Energy and Heat
Summary 16.1 Thermal Energy and Matter Heat flows spontaneously from hot objects to cold objects. • Heat is the transfer of thermal energy from one object to another because of a temperature difference. Temperature is

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related to the average kinetic energy of the particles in

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the transfer of thermal energy from one object to another because of a difference in temperature. temperature. a measure of how hot or cold an object

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is compared to a reference point. absolute zero. a temperature of 0 kelvins. thermal expansion. the increase in volume of a material when its temperature increases.

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Thermal Energy and Heat Work and Heat
Heat is the transfer of thermal energy from one object to another because of a difference in temperature. Heat flows spontaneously from hot objects to cold objects. 16.1 Thermal ...

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